Table 3. A summary of transmission reactions producing only DSC residues

We give fourteen different reaction groups, each exhibiting different numbers of available reactions of various types. The coincidence systems belonging to the reaction groups are also listed below the group descriptions, with the appropriate values of c^2/a^2 given in parentheses.

Reaction type	Number of possible reactions
(a) Reaction group 1	
Perfect → perfect	18
Perfect -> partial	0
Partial → perfect	0
Partial → partial	2
Total	20

(b) Reaction group 2	
Perfect → perfect	18
Perfect → partial	6
Partial → perfect	6
Partial → partial	2
Total	32
$(\frac{12}{5})$: Σ 12, Σ 24, Σ 36, Σ 48.	
(c) Reaction group 3	
Perfect → perfect	18
Perfect -> partial	6
Partial → perfect	6
Partial → partial	594
Total	624
$(\frac{12}{5})$: 516 , 532 , 544 .	

transmission of lattice partial dislocations in h.c.p. crystals, although, for certain values of $(c/a)^2$, $\Sigma = 2N$ and $\Sigma = 3N$ cases may fall into some particular reaction group, as illustrated by the results presented in Table 3.

As for the case of dislocation absorption, transmission or transformation reactions involving one or more lattice partials with site-switching Burgers vectors always produce PDSC dislocations in the grainboundary plane.

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$(Ni,Mg)_{4n+6}Ge_{2n+1}O_{8(n+1)}$, a New Structural Family Related to Olivine and Spinel

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Abstract

A new structural family, $(Ni,Mg)_{4n+6}Ge_{2n+1}O_{8(n+1)}$ $(n = 1, 2, 3, 4, \infty)$, has been identified at atmospheric pressure in the NiO-MgO-GeO₂ system. Its building principle is based on regular intergrowth of n (001) olivine layers (*Pnma* setting) with one {111} cationdeficient rock-salt layer. Alternatively, individual structures are shown to contain spinel elements, the proportion of which decreases with increasing values of *n*. This new family therefore provides a structural transition between the olivine $(n = \infty \text{ end-member})$ and spinel structure types and may be relevant to the high-pressure olivine \rightarrow spinel transformation.

1. Introduction

During the course of a re-investigation of the NiO- $MgO-GeO_2$ system at atmospheric pressure, a new

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compound, $(Ni,Mg)_{10}Ge_3O_{16}$, has recently been identified and characterized by powder X-ray and electron diffraction (Barbier, 1987). Its crystal structure has been described as a 3:1 intergrowth of {111} layers of the spinel and rock-salt structures or, alternatively, as a 1:1 intergrowth of (001) olivine layers (for the *Pnma* setting of the olivine unit cell) and {111} layers of a cation-deficient rock-salt structure (*cf.* Fig. 3*a*).

The latter description shows that the $(Ni,Mg)_{10}Ge_3O_{16}$ structure is very simply related to that of $Mg_{14}Ge_5O_{24}$ determined earlier by Von Dreele, Bless, Kostiner & Hughes (1970) and corresponding to a 2:1 intergrowth of olivine layers and cation-deficient rock-salt layers. This structural relationship suggests the existence of an homologous series based on a variable ratio of olivine (ol) and rock-salt (d) layers with the general formula:

$$M_{4n+6}\text{Ge}_{2n+1}\text{O}_{8(n+1)} = n M_4\text{Ge}_2\text{O}_8(\text{ol}) + M_6\text{GeO}_8(\text{d})$$

where the members n = 1 and n = 2 correspond to the $(Ni,Mg)_{10}Ge_3O_{16}$ and $Mg_{14}Ge_5O_{24}$ structures respectively.

The present paper describes the building principle of this structural family and reports observations by high-resolution electron microscopy (HREM) of some of its members co-existing in the NiO-MgO-GeO₂ system at atmospheric pressure.

2. Description of the $(Ni,Mg)_{4n+6}Ge_{2n+1}O_{8(n+1)}$ family

2.1. Building principle: relation to the olivine structure

The crystal structure of $Mg_{14}Ge_5O_{24}$ (Von Dreele *et al.*, 1970) is shown in Fig. 1 projected on (010) (for the *Pcma* setting of the *Pbam* space group): as described by the original authors, the O atoms form a mixed cubic-hexagonal (hc^2) close-packed array



with metal atoms in octahedral (Mg and Ge) and tetrahedral (Ge) coordination. Along the z direction, double olivine layers (ol) with the stoichiometry $2 Mg_4Ge_2O_8$ alternate with single cation-deficient rock-salt layers (d) with the stoichiometry Mg_6GeO_8 . Accordingly, the strucure can be formally described by the equation:

$$Mg_{14}Ge_5O_{24} = 2 Mg_4Ge_2O_8 (ol) + Mg_6GeO_8 (d).$$

[Note that Mg_2GeO_4 itself crystallizes with the olivine structure at normal pressure and temperatures above 1083 K (Dachille & Roy, 1960).]

For comparison, the olivine structure of α -Co₂SiO₄ (Morimoto, Tokonami, Watanabe & Koto, 1974) is depicted in Fig. 2 in the equivalent projection: the relation to the Mg₁₄Ge₅O₂₄ structure is obvious.

The M_{4n+6} Ge_{2n+1}O_{8(n+1)} (M = metal atom) structural family is simply generated from the Mg14Ge5O24 structure by altering the number n of olivine layers intergrown with a single rock-salt layer. The resulting structures are shown in Fig. 3 (idealized to perfect oxygen close-packing) for the values of n = 1, 2, 3 and 4. One end-member of the series is obviously olivine itself, M_2 GeO₄ ($n = \infty$), and the other end-member is M_{10} Ge₃O₁₆ (n = 1) rather than M_6 GeO₈ (n = 0) as the latter structure would not contain any olivine layer. [It is noteworthy, however, that the member n=0 corresponds to the structure of the mineral murdochite, Cu₆PbO₈ (Christ & Clark, 1955).] Although structures with n < 1 (*i.e.* with more rocksalt layers than olivine layers) are geometrically feasible, none has been identified in this study of the NiO-MgO-GeO₂ system (cf. below), and the M_{4n+6} Ge_{2n+1}O_{8(n+1)} series seems to be restricted to $n \ge 1$.



Fig. 1. The $Mg_{14}Ge_5O_{24}$ structure (Von Dreele *et al.*, 1970) projected on (010) (*Pcma* setting) consisting of double olivine layers (ol) alternating, in the z direction, with single cation-deficient rock-salt layers (d). Large, medium and small circles are Ge, Mg and O atoms respectively. Open, filled and dotted circles are at heights 0, 50 and ~±25 (in units of b/100). Note that some Ge atoms are octahedrally coordinated within the d layers.

Fig. 2. The olivine-type structure of α -Co₂SiO₄ (Morimoto *et al.*, 1974) projected on (010) (*Pnma* setting). Large, medium and small circles are Si, Co and O atoms respectively. Open, filled and dotted circles are at heights 25, 75 and ≈ 0 , 50 (in units of b/100). Note the simple relation to the Mg₁₄Ge₅O₂₄ structure (Fig. 1) and the greater distortion of the oxygen packing in the olivine structure.

As seen in Fig. 3, the close-packed oxygen array changes progressively across the $M_{4n+6}Ge_{2n+1}O_{8(n+1)}$ series: from cubic c for n = 1, to mixed cubichexagonal hc^2 , h^2c^2 and h^3c^2 for n = 2, 3 and 4 respectively, to hexagonal h for $n = \infty$ (olivine). The nth member of the series therefore corresponds to the stacking sequence c^2h^{n-1} of oxygen layers with orthorhombic and monoclinic unit cells for even and odd values of n respectively. Assuming a perfect





Fig. 3. Schematic drawings of the (010) projections of the ideal structures of the $M_{4n+6}Ge_{2n+1}O_{8(n+1)}$ series: (a) M1 (n=1), (b) M2 (n=2), (c) M3 (n=3), and (d) M4 (n=4). A monoclinic pseudocell and the true rhombohedral cell have both been outlined for the M1 structure (cf. Barbier, 1987). Large and small circles are Ge and (Ni,Mg) atoms respectively, O atoms have been omitted. Open, filled and dotted circles are at heights 0, 50 and ± 25 (in units of b/100). Olivine (ol) and rock-salt (d) layers have been indicated. Spinel blocks (3×3 octahedra in M2, M3 and M4) and slabs (in M1) are stippled. Compare with Fig. 4.

close-packing of O atoms with a shortest O···O distance of 2.98 Å, ideal cell parameters can be calculated for any member of the series: they are listed in Table 1 for the structures corresponding to n = 1, 2, 3 and 4. A good agreement is found with the experimental data available for phases in the NiO-MgO-GeO₂ system (Table 1), taking into account that the cell dimensions of a particular structure type decrease with increasing Ni/Ni+Mg ratio (cf. Ringwood, 1961; Navrotsky, 1973).

2.2. Relation to the spinel structure

As described elsewhere (Barbier, 1987), the (Ni, Mg)₁₀Ge₃O₁₆ structure is also simply related to spinel: it consists of an intergrowth of three {111} spinel layers with one {111} rock-salt layer, stacked along the [111] direction of the rhombohedral unit cell (*cf.* Fig. 3*a*). [For comparison, the spinel structure of γ -Fe₂SiO₄ (Marumo, Isobe & Akimoto, 1977) is shown in Fig. 4.] The spinel slabs (S) and rock-salt



Fig. 3 (cont.)

layers (R) in Fig. 3(a) have the stoichiometries $M_7\text{Ge}_2\text{O}_{12}$ and $M_3\text{GeO}_4$ respectively, so that formally, $M_{10}\text{Ge}_3\text{O}_{16} = M_7\text{Ge}_2\text{O}_{12}$ (spinel) + $M_3\text{GeO}_4$ (rock salt). [Note that in the present description, the rock-salt layers are stoichiometric, unlike the cation-deficient d layers in the previous description based on olivine.]

Spinel elements are also readily recognized in the crystal structures of the other members of the $M_{4n+6}Ge_{2n+1}O_{8(n+1)}$ series (Figs. 3b-d). However, the spinel slabs S present in the $M_{10}Ge_3O_{16}$ structure are now broken up into blocks (3×3 octahedra in size) intergrown with smaller rock-salt blocks (3×1 octahedra in size) and olivine-type layers (for $n \ge 3$). As *n* increases, the proportion of spinel blocks decreases while that of olivine layers increases so that the $M_{4n+6}Ge_{2n+1}O_{8(n+1)}$ series represents a transition between spinel-like and olivine-like structures.

The compound $(Ni,Mg)_{10}Ge_3O_{16}$ has been observed experimentally to be unstable at high temperature (at normal pressure) and to decompose according to the reaction (Barbier, 1987):

$$3(Ni,Mg)_{10}Ge_{3}O_{16} \rightarrow (Ni,Mg)_{22}Ge_{9}O_{40} + 8(Ni,Mg)O$$

(n = 1) (n = 4)

Based on the above descriptions of the crystal structures in terms of spinel and olivine elements, this decomposition reaction is equivalent to the well-known high-temperature spinel \rightarrow olivine transformation occurring, for instance, in Mg₂GeO₄ (Dachille & Roy, 1960).



Fig. 4. The spinel structure of γ -Fe₂SiO₄ (Marumo *et al.*, 1977) projected on (100)_t = (110)_c. [The subscripts *c* and *t* refer to the conventional face-centred cubic unit cell and to the smaller body-centred tetragonal cell related to *c* by the matrix $(\frac{1}{2}20/-\frac{1}{2}20/001)$.] Large, medium and small circles are Si, Fe and O atoms respectively. Heights of metal atoms are expressed in 1/8 of the *a_t* repeat and those of O atoms in 1/100. One slab (S) identical to those in Fig. 3(*a*) has been indicated.

3. Observations of M_{4n+6} Ge_{2n+1}O_{8(n+1)} phases in the NiO-MgO-GeO₂ system

Most of the observations reported here have been made on samples synthesized during the investigation of the $(Ni,Mg)_{10}Ge_3O_{16}$ phase (Barbier, 1987) without attempting to determine detailed phase relations in the NiO-MgO-GeO₂ system.

Polycrystalline samples were prepared by hightemperature sintering of high-purity (99.99%) NiO, MgO and GeO_2 powders. The final heat treatment consisted in a heating period of two days at 1673 K in a sealed platinum capsule (to prevent loss of GeO_2) by evaporation) followed by an air quench. Most products were multiphase assemblages and, as expected from the cell parameters in Table 1, a severe overlapping of lines occurred in the powder X-ray diffraction patterns. Accordingly, electron diffraction combined with high-resolution electron microscopy (HREM) proved to be the most suitable technique for phase identification. For this purpose, finely crushed grains were deposited on a holev carbon film supported by a copper grid, and were examined with a Hitachi 800 electron microscope operating at 200 kV and equipped with a tilt-rotation side-entry specimen holder. [A few higher-resolution images (Figs. 8 and 12) have been recorded with a JEOL 200CX electron microscope.]

Some preliminary single-crystal-growth experiments were also performed using a flux technique (PbO flux, slow cooling at 2-4 K h⁻¹ from 1573 to 1173 K). Sizeable crystals were obtained for the phases M1 (n=1), M2 (n=2) and (Ni,Mg)₂GeO₄ olivine ($n=\infty$) (the M1 crystals, however, being invariably multiply twinned).

A schematic phase diagram for the (Ni,Mg)O-rich region of the NiO-MgO-GeO₂ system is shown in Fig. 5. It is based on earlier data published by Ringwood (1961) and Navrotsky (1973) but has been modified in order to include the M1 phase,





 $(Ni_xMg_{1-x})_{10}Ge_3O_{16}[x \approx 0.4-0.5 \text{ (Barbier, 1987)]}, as well as the proposed M2, M3 and M4 phases. The composition range indicated for the M2 phase has been derived from electron microprobe analysis of two flux-grown single crystals, giving an atomic ratio Mg/Ni \approx 4.8. [As indicated in Fig. 5, this range possibly extends toward higher Ni contents.] No data have been obtained concerning the composition ranges of the M3 and M4 phases due to the lack of single-phase products and to unsuccessful attempts to grow single crystals.$

The samples investigated by electron diffraction/microscopy correspond to the three bulk compositions marked a, b and c in the phase diagram of Fig. 5 and to the reaction products from the hightemperature decomposition of the M1 phase. As noted earlier, these samples were multiphase assemblages often consisting, on a microscopic scale, of



Fig. 6. Electron diffraction patterns recorded from single-phase M1, M2 and M4 grains along equivalent crystallographic directions, *i.e.* [1010]_{M1} (hexagonal indexing) and [010]_{M2,M4} (cf. Fig. 3).

complex intergrowths of various phases of the $M_{4n+6}Ge_{2n+1}O_{8(n+1)}$ series. However, single-phase grains of the M1, M2 and M4 phases were not uncommon and Fig. 6 shows the corresponding electron diffraction patterns recorded along equivalent directions, *i.e.* $[1010]_{M1} || [010]_{M2,M4}$ [for the hexagonal unit cell of the M1 phase; note also the orientation relationship (0112)_{M1} (001)_{M2,M4} which is consistent with the structure models of Fig. 3]. The strong 006 (M2) and 0010 (M4) reflections show the six- and ten-layer character of the corresponding structures and the relations $d_{100}(M2) = d_{100}(M4)$ and $d_{024}(M1) = d_{006}(M2) = d_{0010}(M4)$ are consistent with the cell parameters listed in Table 1. [Note that, in the M1 pattern, the $(hkl, -h+k+l \neq 3n)$ reflections are absent as expected from the rhombohedral symmetry, whereas in the M2 and M4 patterns, the forbidden reflections of the Pcma space group, i.e. $(hk0, h \neq 2n)$ and $(0kl, l \neq 2n)$, appear via multiple diffraction.]

As expected from the phase diagram, the sample with bulk composition a gave a mixture of the M1 and M2 phases mainly. They often formed disordered intergrowths as shown in Fig. 7: M1 slabs of various widths, characterized by the 110° monoclinic cell angle and the 9.5 Å spacing (= d_{003} rhombohedral cell or d_{100} monoclinic cell), occur in an M2 matrix with a c repeat of 14.4 Å. All phase boundaries are parallel to (001) planes which is again consistent with the structure models in Fig. 3. Note the presence of several M3-type faults in the left-hand part of Fig. 7 corresponding to a c repeat of ~19 Å. Another faulted M2 grain from the same sample is shown in Fig. 8 containing a narrow slab (about 120 Å wide) of the M3 phase.

Products from the bulk compositions b and c contained mixtures of the M2 and M4 phases along with small amounts of the M3 phase. Figs. 9 and 10 show



Fig. 7. Disordered intergrowth of the M1 and M2 phases observed in a sample with bulk composition a in Fig. 5. The indices in the diffraction pattern refer to the M2 unit cell. Intergrowth takes place on (001) planes as expected from the structural models. Note several M3-type faults with a c repeat of 19 Å.



Fig. 8. Tilted image of a faulted M2 grain containing a narrow slab (about 120 Å wide) of the M3 phase. The latter was only observed as faults or thin slabs of this kind in all the samples examined.



Fig. 9. [010] Zone-axis image of a grain containing M4 as the major phase (*cf.* diffraction pattern) along with a small amount of the M2 phase. Several 19 Å faults are also visible.

Table 1. Cell parameters for some $(Ni,Mg)_{4n+6}Ge_{2n+1}O_{8(n+1)}$ phases

Ideal cell parameters, calculated for a perfect close-packing of G atoms with a shortest $O \cdots O$ distance of 2.98 Å, are given in parentheses.

Compound	a(Å)	b(Å)	c(Å)	β (°)	Reference
Ni ₄ Mg ₆ Ge ₃ O ₁₆ *	10-121 (10-32)	5·886 (5·96)	10·201 (10·32)	109-6 (109-47)	Barbier (1987)
Mg14Ge3O24	10-219 (10-32)	5-944 (5-96)	14-512 (14-60)		Von Dreele et al. (1970)
(Ni ₀₋₁₇ Mg ₀₋₈₃) ₁₄ Ge ₅ O ₂₄ ‡ (Ni ₄ Mg) ₂₂ Ge ₉ O ₄₀	10-18 10-162	5-92 5-914	14-43 24-091		This work This work
(Ni,Mg) ₁₈ Ge ₇ O ₃₂	(10·32) (10·32)	(5·96) (5·96)	(24·33) (19·76)	(100-0)	Nauratela (1073)
Mg2GeO4 olivine	(10-32)	(5.96)	(4.87)		Navrotsky (1973)

* Parameters for the monoclinic pseudocell. The true cell is rhombohedral with the hexagonal parameters a = 5.887 and c = 28.603 Å (Barbier, 1987).

[†] For the Pcma setting of the original Pbam space group.

[‡] Precession film data from a flux-grown single crystal.

[010] zone-axis images of typical grains consisting of single- and multiphase regions. It is noteworthy that, whereas single-phase M2 and M4 grains were commonly encountered, the M3 phase was only observed as a minor component occurring as faults or thin slabs. The measurement of the monoclinic angle of the M3 unit cell in Fig. 10 gives an approximate value of 96° not too different from the ideal angle of 100° listed in Table 1.

It has been mentioned earlier that the M1 phase decomposes at high temperature into a mixture of M4 and rock-salt phases. [The cell parameters listed in Table 1 for the $(Ni,Mg)_{22}Ge_9O_{40}$ phase have actually been refined from powder X-ray data obtained from this decomposition product.] HREM observations, however, revealed the presence of small amounts of M2 and M3 phases intergrown within the M4 grains (Fig. 11) as well as trace amounts of olivine (Fig. 12). Comparison of the high-resolution image of the M4 phase in Fig. 12 with the drawing of its crystal structure in Fig. 3(d) indicates that the



Fig. 10. Faulted M2 grain from a sample with bulk composition b in Fig. 5. The monoclinic symmetry of the M3 slab is clearly visible with a measured angle of about 96°.



Fig. 11. Disordered intergrowth of the M4, M2 and M3 phases resulting from the high-temperature decomposition of the M1 phase.

double rows of white dots in the image can be associated with the rock-salt d layers, alternating in the z direction with four-layer-thick olivine slabs.

4. Discussion

The present experimental observations confirm the existence of a new structural family, (Ni,Mg)4n+6- $Ge_{2n+1}O_{8(n+1)}$. At least three of its members, M1 (n=1), M2 (n=2) and olivine $(n=\infty)$ are stable phases at atmospheric pressure as indicated by the successful growth of single crystals. The M4 phase may be stable at high temperature only as suggested by the observation of large single-phase grains in the products quenched from 1873 K but the failure to obtain single crystals using a flux-growth technique over the temperature interval 1573-1173 K. The presence of olivine in the high-temperature products (cf. Fig. 12) also suggests that the M4 phase may decompose into a mixture of olivine plus rock salt at still higher temperatures. Finally, the M3 phase is perhaps metastable as it could only be identified as small amounts intergrown with other members of the series.

 $Mg_{14}Ge_{3}O_{24}$ is the only phase of the $M_{4n+6}Ge_{2n+1}O_{8(n+1)}$ family to have been reported in the nickel-free MgO-GeO₂ system. It had been wrongly identified as Mg₄GeO₆ in the phase diagram published by Robbins & Levin (1959) (the true



Fig. 12. High-resolution image of an M4 grain observed in the decomposition products of the M1 phase at high temperature. A narrow slab with the olivine structure is intergrown within the grain. Comparison with the structure model in Fig. 3(d) allows the double rows of white dots to be associated with the rock-salt d layers alternating with olivine slabs.

stoichiometry being determined later by Von Dreele et al., 1970) who also reported that it decomposes directly into a mixture of MgO+Mg₂GeO₄ olivine at about 1768 K. In view of the present results for the NiO-MgO-GeO₂ system, it seems likely that this decomposition involves in fact extra intermediate phases such as the M4 phase, *i.e.*, Mg₂₂Ge₉O₄₀. On the other hand, the existence of the M1 phase (Mg₁₀Ge₃O₁₆) in the MgO-GeO₂ system might be restricted to low temperatures because of the high proportion of spinel elements in its crystal structure: indeed, Mg₂GeO₂ spinel itself is only stable below 1083 K (at normal pressure) and transforms into olivine at higher temperatures (Dachille & Roy, 1960).

A structural family similar (if not identical) to the $M_{4n+6}Ge_{2n+1}O_{8(n+1)}$ series probably exists in the CoO-MgO-GeO₂ system, as Navrotsky (1973) reported several unidentified phases corresponding to a molar fraction $X_{\text{GeO}_2} \simeq 0.24$ (cf. $X_{\text{GeO}_2} = 0.23$, 0.26 and 0.29 for the phases M1, M2 and M4 respectively). The existence of such a series in the Co system would only reinforce the similarity with the Ni system: in both cases, a two-phase gap exists along the spinel-olivine join (Navrotsky, 1973) and, as seen in Fig. 5, the $M_{4n+6}Ge_{2n+1}O_{8(n+1)}$ series could be regarded as filling in this gap, not only in a compositional sense but also in a structural sense (cf. § 2). Another system possibly containing a similar series of phases is FeO-MgO-GeO₂: like Ni₂GeO₄ and Co₂GeO₄, Fe₂GeO₄ crystallizes with the spinel structure at normal pressure [e.g. Wyckoff (1965)] and the FeO-MgO join consists also in a continuous solid solution with the rock-salt structure [e.g. Davies & Navrotsky (1983)].

The smooth structural transition between the olivine and spinel structure types provided by the $M_{4n+6}Ge_{2n+1}O_{8(n+1)}$ series may be relevant to the olivine → spinel high-pressure transformation. Just as the spinelloid phases observed in the NiAl₂O₄-Ni2SiO4 and MgGa2O4-Mg2GeO2 systems (Ma, 1974; Akaogi, Akimoto, Horioka, Takahashi & Horiuchi, 1982; Barbier & Hyde, 1986) have been proposed as possible intermediates in a martensitic-type olivine ≠ spinel transition (Hyde, White, O'Keeffe & Johnson, 1982; Madon & Poirier, 1983), the present series could provide an alternative path for the same transformation. It would, however, correspond to a nucleation and growth mechanism as the diffusion of atomic species (i.e. cations) would be required in order to accommodate the different stoichiometries of the various M_n phases. Such a transformation would lead to the following olivine/spinel orientation relationships: $(001)_{ol} || (\bar{1}11)_{sp}$ and $[010)_{ol} || [110]_{sp} (cf.$ Figs. 2 and 3). These relations are identical to those derived from the model involving the spinelloid family [e.g. Hyde et al. (1982)] and are also consistent with experimental observations [e.g. Hamaya & Akimoto (1982), Boland & Liu (1983)]. It is also

interesting to note that the $M_{4n+6}Ge_{2n+1}O_{8(n+1)}$ series is intermediate between olivine and spinel in terms of density of atom packing: in the MgO-GeO₂ system, the structures of Mg₂GeO₄ spinel, Mg₁₄Ge₅O₂₄ and Mg₂GeO₄ olivine correspond to volumes of 17.58, 18.36 and 19.04 Å³ per O atom respectively [values based on data from Von Dreele *et al.* (1970) and Navrotsky (1973)].

Because germanates are usually regarded as lowerpressure analogues of silicates (e.g. Ringwood, 1975), the existence of the $(Ni,Mg)_{4n+6}Ge_{2n+1}O_{8(n+1)}$ series at atmospheric pressure suggests that a similar structural family may exist at high pressure in silicate systems such as MgO-SiO₂. In that respect, it is interesting to point out that in their investigation by electron microscopy of the olivine \rightarrow spinel transition in Mg₂SiO₄, Boland & Liu (1983) observed olivine crystals heavily faulted on (001) planes (for the Pnma setting of the olivine unit cell). Although these faults, observed at low magnification, have been associated with the olivine $\rightarrow \Omega$ -phase transition [the Ω -phase being the 'linking' spinelloid phase - cf. Hyde et al. (1982)], the same type of faults would also be expected from disordered (001) intergrowths of thin slabs of phases from the $Mg_{4n+6}Si_{2n+1}O_{8(n+1)}$ series (cf. Figs. 7-11). It appears that high-resolution electron microscopy would be required in order to distinguish between, in particular, images of thin slabs of the two linking M1 and Ω structures: both correspond to a cubic close-packing of O atoms but, whereas the Ω -phase contains olivine-type layers only (Hyde et al., 1982), the M1 phase is a 1:1 intergrowth of olivine and rock-salt layers. The only difference between the two structures therefore resides in the cation content and distribution within half of the cubic close-packed layers. Finally, it may be noted that the cation distribution in the structures of the hypothetical $Mg_{4n+6}Si_{2n+1}O_{8(n+1)}$ series would require some Si atoms to be six-coordinated within the d rock-salt layers. Although this condition suggests that these structures could only be stable at relatively high pressures, it may not be prohibitive as very high pressures are indeed involved in these transformations [*e.g.* 220 kbar (22 GPa) in Boland & Liu's experiment].

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$K_x(Nb,W)_{17}O_{47}$ (1 $\leq x \leq$ 2): a New Tunnel Structure Derived from High-Resolution Electron Micrographs

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Abstract

A new tunnel structure of a fully oxidized compound with the general formula $K_x Nb_{8+x} W_{9-x}O_{47}$ ($1 \le x \le 2$) has been deduced from high-resolution electron micrographs. For x = 2 the monoclinic unit-cell parameters are a = 18.882 (5), b = 3.9572 (5), c =12.378 (2) Å, $\beta = 102.93$ (3)°, V = 901.41 Å³, Z = 1, $M_r = 3046.2$ and $D_x = 5.61$ g cm⁻³. The space group is P2/m. The structure has been confirmed by simulated image calculations and by X-ray powder diffraction studies. The polyhedral framework is built up of octahedra and pentagonal columns in such a way that four-, five- and six-sided tunnels are formed. The structure comprises features of both the W₁₈O₄₉ and the tetragonal tungsten bronze (TTB) structure types.

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